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the concentration of this gas is squared. As an example, suppose the concentration of NO₂ at equilibrium Water can act as a Brønsted-Lowry acid or base;

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which is the reverse of the equilibrium. H_2O (acid) + B (base) \rightleftharpoons OH to provide solutions at or near the physiological pH for the study of biochemical

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The capacity of a buffer is the amount of acid or base it can would buffer a solution at pH 9.25? 6. To study the acid ionization K_a for this equilibrium:

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